DISTRIBUTION OF CHLORIDE, PH, RESISTIVITY, AND SULFATE LEVELS IN BACKFILL FOR MSE WALLS AND IMPLICATIONS FOR CORROSION TESTING

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Progress Report BDV 25 TWO 977-03

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PROJECT GOALS

Assure that:

- For chloride, pH, resistivity, and sulfate levels measured in corrosion testing, variability in these levels due to sampling and analytical techniques is much lower than variability within a select backfill stockpile or stratum.
- Corrosion properties of backfill material do not change appreciably over time, especially after emplacement and over the design lifetime of the MSE wall.
- The number of soil samples analyzed prior to acceptance of backfill is appropriate.

PROJECT TASKS

Task 1 Literature Review

Task 2 Databased Trends in MSE Wall Backfill Properties

Task 3 Single-Laboratory Contributions to Method Reproducibility & Proposed Method Improvements

Task 4 Heterogeneity of MSE Wall Backfill

Task 5 Rainfall-Driven Temporal Changes in Backfill Properties

Task 6 Multi-Laboratory Contributions to Method Reproducibility

Task 7 Final Report

08/01/2014

PROJECT PROGRESS BY TASK

Task	10%	20%	30%	40%	50%	60%	70%	80%	90%	100%
1-A										
1-B										
2										
З-А, З-В										
3-C, 3-D										
3-E, 3-F										
3-G, 3-H										
3-I, 3-J										
4										
5										
6-A										
6-B										
7										

Overall Task Progress is 55%

PROJECT PROGRESS BY BILLING



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08/01/2014

RESEARCH HIGHLIGHTS FOR FM5-550 pH IN SOIL AND WATER







For pH, 3% of the samples were below pH 5 or above pH 9; the failure rate was 1% for District 1 and 4% for District 7.

TASK 3 SINGLE-LABORATORY PERFORMANCE

Sample	Ν	Average	St Dev	%RSD	%RE
pH 5 Buffer	5	5.00	0.01	0.23	0.32
pH 9 Buffer	5	9.02	0.00	0.00	-0.22
Santa Fe River Sand	9	7.97	0.06	0.73	2.13
Starvation Hill Sand	13	7.82	0.07	0.90	3.10

TASK 4 HETEROGENEITY OF BACKFILL



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pH 9 Buffer	5	9.02	0.00	0.00	-0.22
Santa Fe River Sand	9	7.97	0.06	0.73	2.13
Starvation Hill Sand	13	7.82	0.07	0.90	3.10
Calhoun Sand	12	4.64	0.21	4.61	18.5
Jahna Sand	12	5.37	0.22	4.11	15.5
Wimauma Sand	12	4.66	0.09	1.87	6.0
Youngquist Sand	12	7.98	0.30	3.81	12.3

TASK 6 MULTI-LABORATORY PERFORMANCE



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TASK 6 MULTI-LABORATORY PERFORMANCE

On The Road With Santa Fe River Sand							
pH Measured by	Ν	Average	St Dev	%RSD	%RE		
USF Laboratory	9	7.97	0.06	0.73	2.13		
USF On-Site Audit	14	8.15	0.15	1.83	6.62		
Inter-Laboratory	15	7.85	0.80	10.1	32.4		

Note: Results for 6 FDOT and 9 commercial laboratories; these results do not include variability from field sampling, transport, and storage of soil.

TASK 6 MULTI-LABORATORY PERFORMANCE

Transported soil at ambient vs cool temperatures Stored under ambient conditions or in the refrigerator Used 100 g or 100 ml of soil Tested "as is" or air dried Used soil from resistivity measurement Waited zero to 30 min or more to test sample Stirred once or up to three times Put soil/water mixture on shaker table for three 10-min intervals Stirred or did not stir sample during measurement Took reading when stable light came on or waited until reading was stable for 1 min Prepared soil water/mixture in a beaker, bottle, or disposable cup Calibrated with one, two, or three buffers Used fresh buffers or re-used buffers for calibration Stored pH electrode in distilled water, tap water, buffer, KCI, or dry Used glass electrode or solid state electrode Refilled glass electrode or used disposable electrode Kept electrode for up to 10 yrs

08/01/2014

TASK 1 LITERATURE REVIEW



Basic parts of a 3-in-1 combination electrode.

MSE Wall Backfill

Problem:

pH measurement is slow, drifting, noisy, non-reproducible, or inaccurate.

Causes:

Electrolyte is contaminated, liquid junction is clogged, glass bulb is damaged, reference element is depleted, wire is broken, or buffers are contaminated.

TASK 1 LITERATURE REVIEW

Sources of electrical potential in a combination electrode; ideally, E2 through E6 are held constant and E1 varies with the hydronium ion concentration in the solution.



MSE Wall Backfill

Problem:

pH measurement is slow, drifting, noisy, non-reproducible, and inaccurate.

Causes:

Low electrical conductivity of sample, differences between low ionic strength solutions and normal ionic strength buffers, change in the liquid junction potential, and absorption of carbon dioxide.

TASK 3 PROPOSED IMPROVEMENTS

Standardize equipment and electrode capabilities

- Develop and include QA/QC for pH electrode
 - Reach a stable soil reading within ~1 min
 - Read buffer pH ±0.05 pH units of buffer concentration
 - Calibrate to a % slope within 90% to 102%
 - Check that offset is within ± 25 mV
- Provide more detail in procedures
 - Calibrate with three fresh buffers
 - Take steps check electrode performance
 - Provide steps for cleaning, storing, and filling pH electrode
- Make pH method more robust
 - Air dry and sieve soil?
 - Use an ionic strength adjuster—add 0.1 g KCl to samples?
 - Test more samples?

TASK 3 PROPOSED IMPROVEMENTS



QUESTIONS OR COMMENTS?



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TASK 3 SINGLE-LABORATORY PERFORMANCE

Ruggedne	pH Determination								
Original	Change	1	2	3	4	5	6	7	8
100 ml (ML) soil	100 g (G) soil	ML	ML	ML	ML	G	G	G	G
room temp (RT)	cold (C)	RT	RT	С	С	RT	RT	С	С
30 min wait	no wait	30	0	30	0	30	0	30	0
distilled (DSTL)	deionized (DI)	DSTL	DSTL	DI	DI	DI	DI	DSTL	DSTL
pH probe 1	pH probe 2	1	2	1	2	2	1	2	1
stir gently (G)	no stir (NS)	G	NS	NS	G	G	NS	NS	G
wetted soil	dry soil	WET	DRY	DRY	WET	DRY	WET	WET	DRY
pH Results: 8	7.99	8.01	7.90	8.18	7.94	8.40	8.00	8.31	

TASK 3 SINGLE-LABORATORY PERFORMANCE

Original	Change	Difference	pH is higher if		
30 min wait	no wait	-0.267	measured without delay		
100 ml soil	100 g soil	-0.143	water to soil ratio is higher		
distilled (DSTL) water	deionized (DI) water	-0.027	diluted with deionized water		
room temp (RT) sample	cold sample (C)	-0.012	sample is colder		
stir gently (G) with probe	no stir (NS) with probe	0.027	sample is gently stirred		
soil is wetted overnight	soil is dry	0.102	soil is wetted overnight with 10% water		
pH probe 1	pH probe 2	0.118	measured with probe 1		